

Preparation and Characterization of Amorphous Carbon Nanotube-MoS₂ Nanohybrid

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Abstract: Recently amorphous carbon nanotube have become alternative to the crystalline carbon nanotube due to their easy synthesis procedure. Numerous properties of amorphous CNT can be varied through hybridizing it with several nanomaterial. MoS₂ nanosheets has good optoelectronic and physical property, which makes it suitable as an additive with other material. In this work, amorphous CNT – MoS₂ nanohybrid were synthesized by using hydrothermal method. Amorphous carbon nanotube has been prepared separately through a low temperature synthesis process. Nanohybrids were characterized by using X-ray diffraction, field emission scanning electron microscope, high resolution transmission electron microscope, energy-dispersive X-ray spectroscopy. Band gap of as prepared hybrids were also calculated.

Keywords: amorphous carbon nanotube, MoS₂, nanohybrids.

1. Introduction

In recent days, nanostructure material plays a vital role in the field of nanoscience and nanotechnology. Electronic devices are shrinking day by day as devices are becoming nanometer size. Though silicon is most useful material for semiconductor device fabrication for some good features. Researchers are doing extensive studies for developing other zero, one and two dimensional material, so that they can be useful in modern device fabrication. Among all the nanostructure material, carbon nanotube [1] is most interesting two dimensional material due to its amazing optical, electronic and mechanical properties [2]-[4] which makes them suitable for microelectronics device fabrication. Depending on its crystalline property, it can be classified into two types such as crystalline carbon nanotube [5] and amorphous carbon nanotube [6]. So far, several methods like arc discharge [7], laser ablation [8], chemical vapour deposition[9] have been discovered for preparation of crystalline carbon nanotube. However, these methods are very complicated and require high temperature like above 800°C. On the other hand, amorphous carbon nanotube can be synthesized by simple chemical process and it almost require very less temperature such as 225°C. Furthermore, defects and dangling bond in amorphous CNT makes them suitable to react with environment easily and can formed in different nanostructure hybrids. On the other hand,

among metal chalcogenides [10], MoS₂ is very important material due to its fascinating electronic and physical property [11]-[22]. Bulk MoS₂ is an indirect band gap material with band gap of 1.2 eV whereas MoS₂ nanosheets is a direct band gap material with band gap of 1.8 eV. Owing to its fascinating properties, it is widely used in energy storage [23], solar cells [24], wearable devices [25], catalysis [26], double-layer capacitors [27], lubricants [28], photo emitting devices [29], hydrogen storage [30]. MoS₂ is used to enhance the piezoelectricity of polymer based Nano generator [31-41]. For synthesis of MoS₂ Nano sheets, several methods like liquid exfoliation [42], wet chemical reaction [43], hydrothermal process [44], chemical vapour deposition[45] etc. have been discovered. Here in this paper, amorphous carbon nanotube – MoS₂ nanohybrid synthesis procedure and its characterisation has been done.

2. Material and Methods

Ferrocene ((C₅H₅)₂Fe, Merck), ammonium chloride (NH₄Cl) and hydrochloric acid (HCl) are the main material used in this process. For synthesis of amorphous CNT, first ferrocene and ammonium chloride were taken in 1:2 weight ratio. Then weighted material placed in a mortar and grinded very well. After that, the mixture were placed in a quartz beaker and kept it in an oven with 225°C for 30 minutes. Then the black product was washed with DI-water and dilute HCL repeatedly for removing the impurity present in amorphous CNT. Then final product was kept in an oven with 60°C for 24 hrs.

For synthesis of amorphous CNT-MoS₂ nanohybrid, 2 mmol of sodium molybdate dihydrate were dissolved together in 50 ml deionized (DI) water for 2 min. Then, 10 mmol of L-cysteine was added to the solution followed by sonication for 30 minutes. Afterthat, 10 M HCl was added to the solution drop by drop until the pH value of the solution was become less than 1. This solutions were stirred for 30 minutes for making a homogeneous solutions. Then the solutions were transferred into a teflon lined stainless steel autoclave and 0.5 g of as prepared amorphous CNT were poured into it. Solutions contained autoclave were kept in a hot air oven at 200°C for

Table 1
 Compositional analysis of amorphous CNT – MoS₂ hybrid sample from EDX study

Element Line	Net Counts	Net Counts Error	Int. Cps/nA	Int. Error	Weight %	Weight % Error	Atom %	Atom % Error	Formula
C K	1025	+/- 96	---	---	25.88	+/- 0.81	48.00	+/- 4.50	C
O K	1910	+/- 183	---	---	21.53	+/- 0.69	29.97	+/- 2.87	O
S K	11560	+/- 1308	---	---	21.23	+/- 0.80	14.75	+/- 1.67	S
S L	0	+/- 51	---	---	---	---	---	---	---
Mo L	12889	+/- 1914	---	---	31.36	+/- 1.55	7.28	+/- 1.08	Mo
Mo M	142	+/- 60	---	---	---	---	---	---	---
Total					100.00		100.00		

25hrs. Finally, the heated solutions were kept in room temperature for a definite time and washed with di-water and ethanol repeatedly. Finally, the product was dried at 60°C for 16 hrs.

3. Characterization and results

Amorphous CNT – MoS₂ nanohybrids has been characterized by field emission scanning electron microscope (JEOL 6340F FEG-SEM), high resolution transmission electron microscope (JEOL-JEM 2100), X-ray diffraction (XRD, Bruker), energy-dispersive X-ray spectroscopy.

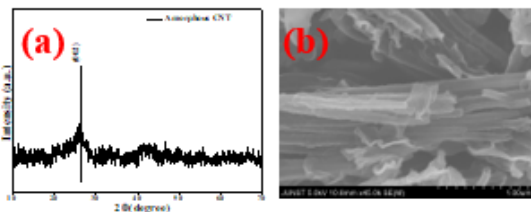


Fig. 1. (a) XRD and (b) FE-SEM image of amorphous carbon nanotube

Figure 1(a) shows the FE-SEM image of amorphous carbon nanotube sample. From there it can be observed the formation of carbon nanotube. Figure 1(b) shows the XRD pattern taken using Cu K α radiation (wavelength $\lambda = 0.15418$ nm) with normal θ - 2θ scanning in the range between 10 - 70 ° of as synthesized amorphous carbon nanotube. Sample shows one peak centering around $2\theta \approx 26.5^\circ$ which is the for the (002) plane of 2 dimensional graphitic phase of carbon. However, the peaks is not prominent which means it has poor crystallinity.

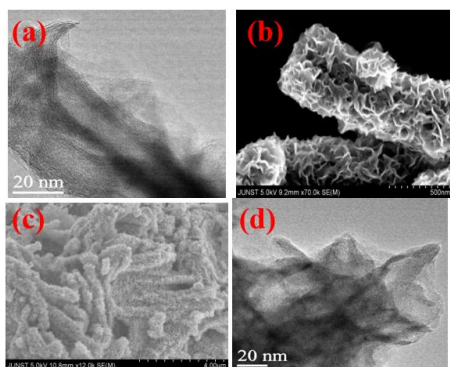


Fig. 2. (a and b) FE-SEM and (c and d) HRTEM images of amorphous CNT – MoS₂ nanohybrid

From FE-SEM and HRTEM images, it is clearly observed

the formation of MoS₂ nanosheets decorated amorphous carbon nanotube. Compositional analysis of as prepared nanohybrids were done by EDAX analysis. From there, it is observed that atom ratio between Mo and S is about 1:2.026, which is very close to the stoichiometric MoS₂. Apart from Mo and S, there are also present carbon element which is corresponds to amorphous carbon nanotube.

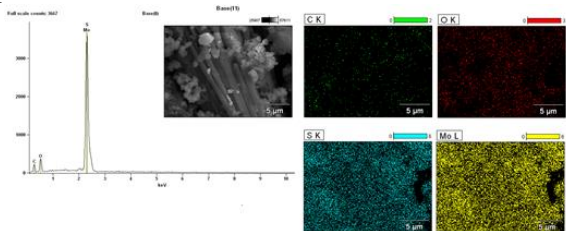


Fig. 3. EDX spectra of ACNT – MoS₂ hybrid sample with corresponding elemental mapping

X-ray diffraction (XRD) is a rapid analytical technique primarily used for phase identification of a crystalline material and can provide information on unit cell dimensions. Here, XRD pattern of amorphous CNT-MoS₂ nanohybrid have been taken using Cu K α radiation (wavelength $\lambda = 0.15418$ nm) with normal θ - 2θ scanning in the range between 10 - 80°. There is present two hump at in the XRD pattern of ACNT-MoS₂ sample at $2\theta = 33^\circ$ and 56.63° which is corresponds to the (100) and (110) plane of MoS₂. On the other side, band gap is a major factor for determining optoelectronic properties of any nanomaterial. Specially the electrical conductivity can be figure out from band gap study. Here band gap of as synthesized sample has been determined by using tauc plot and its estimated value is 2.6 eV.

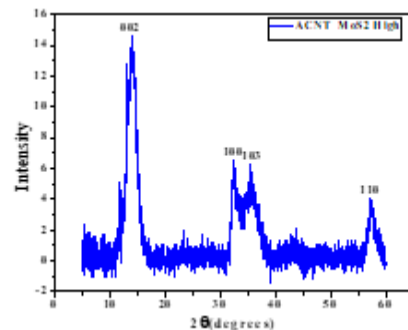


Fig. 4. XRD pattern amorphous CNT – MoS₂ nanohybrid

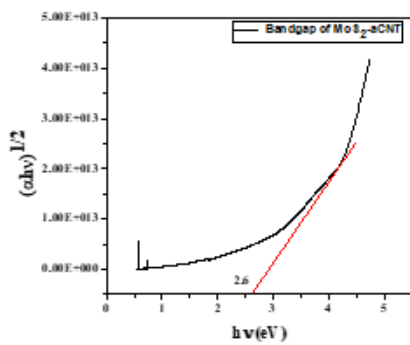


Fig. 5. Band gap plot of amorphous CNT – MoS2 nanohybrid

4. Conclusion

Synthesis procedure of amorphous carbon nanotube are quiet easy and it can be used as a replacement of crystalline carbon nanotube in several application by growing it with some other nanomaterial. Amorphous-CNTs have been successfully synthesized via a relatively simple technique. The detection of hydroxyl groups implies that the walls of the nanotubes are amorphous, with the introduction of numerous defects. Here amorphous CNT- MoS2 nanohybrids has been synthesized by using hydrothermal process. These composite may be useful for removing toxic dye from water and supercapacitor application.

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